

<b>Core</b>	<b>95 hours</b>
<b>Topic 1: Stoichiometric relationships</b>	<b>13.5</b>
1.1 Introduction to the particulate nature of matter and chemical change	
1.2 The mole concept	
1.3 Reacting masses and volumes	
<b>Topic 2: Atomic structure</b>	<b>6</b>
2.1 The nuclear atom	
2.2 Electron configuration	
<b>Topic 3: Periodicity</b>	<b>6</b>
3.1 Periodic table	
3.2 Periodic trends	
<b>Topic 4: Chemical bonding and structure</b>	<b>13.5</b>
4.1 Ionic bonding and structure	
4.2 Covalent bonding	
4.3 Covalent structures	
4.4 Intermolecular forces	
4.5 Metallic bonding	
<b>Topic 5: Energetics/thermochemistry</b>	<b>9</b>
5.1 Measuring energy changes	
5.2 Hess's Law	
5.3 Bond enthalpies	
<b>Topic 6: Chemical kinetics</b>	<b>7</b>
6.1 Collision theory and rates of reaction	
<b>Topic 7: Equilibrium</b>	<b>4.5</b>
7.1 Equilibrium	

<b>Topic 8: Acids and bases</b>	<b>6.5</b>
8.1 Theories of acids and bases	
8.2 Properties of acids and bases	
8.3 The pH scale	
8.4 Strong and weak acids and bases	
8.5 Acid deposition	
<b>Topic 9: Redox processes</b>	<b>8</b>
9.1 Oxidation and reduction	
9.2 Electrochemical cells	
<b>Topic 10: Organic chemistry</b>	<b>11</b>
10.1 Fundamentals of organic chemistry	
10.2 Functional group chemistry	
<b>Topic 11: Measurement and data processing</b>	<b>10</b>
11.1 Uncertainties and errors in measurement and results	
11.2 Graphical techniques	
11.3 Spectroscopic identification of organic compounds	
<b>Additional higher level (AHL)</b>	<b>60 hours</b>
<b>Topic 12: Atomic structure</b>	<b>2</b>
12.1 Electrons in atoms	
<b>Topic 13: The periodic table—the transition metals</b>	<b>4</b>
13.1 First-row d-block elements	
13.2 Coloured complexes	
<b>Topic 14: Chemical bonding and structure</b>	<b>7</b>
14.1 Covalent bonding and electron domain and molecular geometries	
14.2 Hybridization	
<b>Topic 15: Energetics/thermochemistry</b>	<b>7</b>
15.1 Energy cycles	
15.2 Entropy and spontaneity	

<b>Topic 16: Chemical kinetics</b>	<b>6</b>
16.1 Rate expression and reaction mechanism	
16.2 Activation energy	
<b>Topic 17: Equilibrium</b>	<b>4</b>
17.1 The equilibrium law	
<b>Topic 18: Acids and bases</b>	<b>10</b>
18.1 Lewis acids and bases	
18.2 Calculations involving acids and bases	
18.3 pH curves	
<b>Topic 19: Redox processes</b>	<b>6</b>
19.1 Electrochemical cells	
<b>Topic 20: Organic chemistry</b>	<b>12</b>
20.1 Types of organic reactions	
20.2 Synthetic routes	
20.3 Stereoisomerism	
<b>Topic 21: Measurement and analysis</b>	<b>2</b>
21.1 Spectroscopic identification of organic compounds	